intricate feature of myoglobin, which is essential for respiration. The quality of the data depends on the ability to dissociate, non-destructively, a substantial fraction of unit cells. It turned out to be beneficial to increase the laser pulse length from 150 fs to 2 ps to maximise the dissociation yield. This trick increased the number of "correctly excited" unit cells by 10, which gave a 10-fold improvement in the signal to noise. This effect is due to the presence of short-lived, high-absorption electronic states that tend to obstruct the entry of the laser pulse.

The experiments on excited molecules in liquids are reported subsequently by A. Plech et al., who studied the excited-state structure of iodine in liquid CCl$_4$. The initial aim was to extract the diffracted signal from excited iodine, determine the change in the I$_2$ bond length and the degree of dissociation. It came as a surprise, therefore, to see that a local change in structure around iodine can induce a global change in structure in the bulk liquid, driven by the flow of energy from the solute to the liquid. This forces the liquid to expand. The expansion shows up after 10 ns and is over after 100 ns. This pattern is quite general in photo-excited liquids and we now classify the liquid structure into a pre-expansion, an expansion and post-expansion phase. The expansion amplitude, defined by the change in distance between neighbouring solvent molecules, is typically a fraction of a milliangstrom, but is shows up since it integrates all the solvent molecules in the sample. From the expansion amplitude, it is possible to check the associated energy input from the laser, which is a way to check if the total energy is conserved.

**Picosecond Crystallography: Myoglobin in Action**

Proteins perform their life-giving function dynamically while interacting with other molecules. The ability to watch a protein as it responds to stimuli will help researchers solve many mysteries regarding how it achieves its function. A new step toward that goal was recently taken by a team of scientists working on the time-resolved beamline **ID09B**, were the first ever movie of a protein executing its function with sub-nanosecond time resolution was produced. The system studied by this team, the oxygen storage protein myoglobin, is similar to hemoglobin but is found in muscle cells rather than blood cells. It supplies oxygen to cells when blood fails to deliver it fast enough (for example, when circulation is blocked during muscle contraction). The aim of the experiment was to identify pathways for oxygen diffusion within the protein and to see how oxygen is finally released into the surrounding solvent. By using CO as a substitute for oxygen, they took advantage of the photosensitivity of the Fe-CO bond, allowing them to trigger ligand release synchronously throughout a crystal by a short laser flash. The mutant L29F of myoglobin [1], in which a leucine residue is replaced by a phenylalanin, proved to be an interesting test case because substantial protein conformational changes take place during the first few hundred picoseconds after laser photolysis.

This experiment makes use of the time structure of synchrotron radiation, which consists of 50 to 150 ps long X-ray pulses. The **ID09B** beamline is equipped with a high-speed chopper capable of isolating single X-ray pulses from the synchrotron pulse train and houses an ultrashort pulsed laser that is synchronized to the X-ray pulses. Time-resolved crystallographic data were acquired using the “pump-probe” technique: the sample was first hit by a laser pulse to trigger the reaction (pump) and then a time-delayed X-ray pulse, whose diffraction pattern is recorded by an area detector. A cooling stream allows the flash photolysis to be repeated at 3.3 Hz. The orientation of the sample is fixed during each exposure; the goniometer spindle rotated between exposures. The 200 µm crystal is mounted in a sealed X-ray capillary.

Fig. 1: Experimental setup used to acquire picosecond time-resolved X-ray diffraction data. The crystal is photolyzed by a laser pulse and then probed by a time delayed X-ray pulse, whose diffraction pattern is recorded by an area detector. A cooling stream allows the flash photolysis to be repeated at 3.3 Hz. The orientation of the sample is fixed during each exposure; the goniometer spindle rotated between exposures. The 200 µm crystal is mounted in a sealed X-ray capillary.

The earliest map, 100 ps after photolysis ([Figure 3a](#fig3a)), shows that the protein responds promptly to CO dissociation from the iron, revealing distortions spanning across the entire
molecule. The time series (Figure 3b-g) suggests a mechanism responsible for the rapid CO expulsion that is characteristic of the L29F mutant. To make room for the free CO, the mutant side chain Phe29 is pushed aside together with the His64. The strain energy stored in this configuration helps drive CO out of its primary docking site on a timescale a thousand-fold faster than in wild-type myoglobin. The CO does not immediately escape into the surrounding water from this site, but is still found inside the protein after three microseconds as it migrates through an interconnected web of hydrophobic cavities and seeks an escape channel.

Given the level of structural detail obtained with picosecond X-ray crystallography, there is hope that it will contribute to our understanding of protein function at an atomic level.

References

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